

# Lithium density kg m<sup>3</sup>

Densities of some common metals, metallic elements and alloys - aluminum, bronze, copper, iron and more.

Density Properties Elastic Properties Hardnesses Heat and Conduction Optical Properties Density of solid: 535 kg/m<sup>3</sup>; Molar volume: 13.02 cm<sup>3</sup> See more on webelements aqua-calc Lithium weight to volume conversion - Aqua-Calc Lithium weighs 0.534 gram per cubic centimeter or 534 kilogram per cubic meter, i.e. density of lithium is equal to 534 kg/m<sup>3</sup>; at 20°C (68°F or 293.15K) at standard atmospheric pressure.

The density, or more precisely, the volumetric mass density of lithium is 535 kg/m<sup>3</sup>. This means that the mass of 1 m<sup>3</sup> lithium is 535 kg. The weight (on earth) of 1 m<sup>3</sup> lithium is 5248,35 N or 5,24835 kN. ...

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Different metals have different densities, and the density difference between some metals is huge. For example, the most dense metal is Osmium (Os), with a density of 22.59 g/cm<sup>3</sup>, which is 42 times ...

Everything has mass and volume, and therefore density. So it is expected that the smallest piece of matter, like an atom of Lithium has it too. But what is the density of an atom of Li? ...

Density of Li (Lithium) is 534 kg/m<sup>3</sup> [ Convert 534 kg/m<sup>3</sup> to different units ] Find density of ...

>Mass density of Lithium is 534 kg/m<sup>3</sup>. Lithium specific gravity, mass- and volume calculator.

It has an estimated density of 40.7 x 10<sup>3</sup> kg/m<sup>3</sup>. The density of Hassium results from its high atomic weight and from the significant decrease in ionic radii of the elements in the lanthanide ...

Heatscape picture representing the density of the periodic table elements. Heatscape representing the molar volume of the periodic table elements.

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